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phase energy (Eph) -energy stored in system due to arrangement of particles that exert attractions on one another. (energy account involved when phase changes occur) -Attractions result in decrease of energy of a system of particles. -As particles become more tightly bound, their phase energy is lowered. -Solids possess lowest phase energy, and liquids possess more because the particles move freer.

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Describe the ways information is like energy, according to your reading. Money accounts are another metaphor that can help us understand energy storage and transfer. Describe the ways money accounts are like energy. We describe three storage ||accounts|| to understand the changes we see in chemistry.

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84 Study Guide for An Introduction to Chemistry Section Goals and Introductions Section 7.1 Energy Goals To introduce the terms energy, kinetic energy, and potential energy. To introduce the Law of Conservation of Energy. To describe the relationships between stability, capacity to do work, and potential energy.

[Chapter 7 Energy and Chemical Reactions](#)

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32 Study Guide for An Introduction to Chemistry Exercise 3.3 - Cations and Anions: Identify each of the following as a cation or an anion, and determine the charge on each. (Obj 22) a. magnesium atom with 12 protons and 10 electrons (+12) + (||10) = +2 This is a +2 cation. b. fluorine atom with 9 protons and 10 electrons

[Chapter 3 The Structure of Matter and the Chemical Elements](#)

1. Energy is the ability to do work or cause change and occurs in many forms. 2. The two main types of energy are potential energy and kinetic energy. 3. Define Potential Energy: Potential energy is stored energy or energy at rest. Potential energy can be increased or decreased due to position or condition. 4.

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The modern science of chemistry began during the eighteenth century, when several brilliant natural philosophers classified the products of decomposition into a small number of fundamental substances. For example, in 1774, the Englishman Joseph Priestley discovered that when the red powder mercuric oxide was heated, it decomposed to liquid metal mercury and to a colorless gas capable of ...

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