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This excellent new text dispels the fear that the word electrochemistry commonly instills in chemistry students. Throughout the mathematical content has been left to a minimum for clarity, whilst retaining the important necessary physical insight.

Another winning primer! This new addition to the popular series provides a basic introduction to equilibrium electrochemistry, focusing on electrode potentials and their applications. It builds on a knowledge of elementary thermodynamics giving the student an appreciation of the origin of electrode potentials and shows how these are used to deduce a wealth of chemically important information and data such as equilibrium constants, the free energy, enthalpy and entropy changes of chemical reactions, activity coefficients, the selective sensing of ions. It is mathematically simple, the emphasis throughout is on understanding the foundations of the subject and how it may be used to study problems of chemical interest.

This textbook provides a framework of the key concepts involved in electrochemical kinetics. A wide range of modern electrochemical techniques and applications are discussed. The mathematical content has been minimized for clarity, while retaining the important results necessary for physical insight. A substantial series of examples and illustrations is taken from the recent research literature to explore the potential applications of electrochemical techniques.

The renowned Oxford Chemistry Primers series, which provides focused introductions to a range of important topics in chemistry, has been refreshed and updated to suit the needs of today's students, lecturers, and postgraduate researchers. The rigorous, yet accessible, treatment of each subject area is ideal for those wanting a primer in a given topic to prepare them for more advanced study or research. Moreover, cutting-edge examples and applications throughout the texts show the relevance of the chemistry being described to current research and industry. The learning features provided, including questions at the end of every chapter and online multiple-choice questions, encourage active learning and promote understanding. Furthermore, frequent diagrams, margin notes, further reading, and glossary definitions all help to enhance a student's understanding of these essential areas of chemistry. This brand new addition to the series provides the most accessible first introduction to electrochemistry, combining explanation of the fundamental concepts with practical examples of how they are applied in a range of real-world situations.

Understanding the mechanisms of the reactions at transition metal sites is a key component in designing synthetic methods, developing industrial homogeneous catalysts, and investigating metalloenzymes. These mechanisms are therefore an essential part of undergraduate chemistry courses. This primer provides a broad-based, systematic guide to the fundamentals of transition-metal mechanistic chemistry, including substitution, electron transfer, and reactions of ligands. It serves as an ideal text for undergraduate students with a foundation in basic inorganic chemistry but who are new to inorganic reaction mechanisms.

Comprehensive text and reference covers all phenomena involving light in semiconductors, emphasizing modern applications in semiconductor lasers, electroluminescence, photodetectors, photoconductors, photoemitters, polarization effects, absorption spectroscopy, more. Numerous problems. 339 illustrations.

This book will give students a thorough grounding in pH and associated equilibria, material absolutely fundamental to the understanding of many aspects of chemistry. It is, in addition, a fresh and modern approach to a topic all too often taught in an out-moded way. This book uses new theoretical developments which have led to more generalized approaches to equilibrium problems; these approaches are often simpler than the approximations which they replace. Acid-base problems are readily addressed in terms of the proton condition, a convenient amalgam of the mass and charge constraints of the chemical system considered. The graphical approach of Bjerrum, Hagg, and Sillen is used to illustrate the orders of magnitude of the concentrations of the various species involved in chemical equilibria. Based on these concentrations, the proton condition can usually be simplified, often leading directly to the value of the pH. In the description of acid-base titrations a general master equation is developed. It provides a continuous and complete description of the entire titration curve, which can then be used for computer-based comparison with experimental data. Graphical estimates of the steepness of titration curves are also developed, from which the practicality of a given titration can be anticipated. Activity effects are described in detail, including their effect on titration curves. The discussion emphasizes the distinction between equilibrium constants and electrometric pH measurements, which are subject to activity corrections, and balance equations and spectroscopic pH measurements, which are not. Finally, an entire chapter is devoted to what the pH meter measures, and to the experimental and theoretical uncertainties involved.

This book is a concise introductory guide to understanding the foundations of electrochemistry. By using simplified classroom-tested methods developed while teaching the subject to engineering students, the author explains in simple language an otherwise complex subject that can be difficult to master for most. It provides readers with an understanding of important electrochemical processes and practical industrial applications, such as electrolysis processes, metal electrowinning, corrosion and analytical applications, and galvanic cells such as batteries, fuel cells, and supercapacitors. This powerful tutorial is a great resource for students, engineers, technicians, and other busy professionals who need to quickly acquire a solid understanding of the science of electrochemistry.

The leading reference on electroencephalography since 1982, Niedermeyer's Electroencephalography is now in its thoroughly updated Sixth Edition. An international group of experts provides comprehensive coverage of the neurophysiologic and technical aspects of EEG, evoked potentials, and magnetoencephalography, as well as the clinical applications of these studies in neonates, infants, children, adults, and older adults. This edition's new lead editor, Donald Schomer, MD, has updated the technical information and added a major new chapter on artifacts. Other highlights include complete coverage of EEG in the intensive care unit and new chapters on integrating other recording devices with EEG; transcranial electrical and magnetic stimulation; EEG/TMS in evaluation of cognitive and mood disorders; and sleep in premature infants, children and adolescents, and the elderly. A companion website includes fully searchable text and image bank.

This is an introduction to the areas of application of electroanalysis, which has an important role with current environmental concerns, both in the laboratory and in the field.

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